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# PURIFICATION AND PARTIAL CHARACTERIZATION OF A PROCARYOTIC GLYCOPROTEIN FROM THE PLASMA MEMBRANE OF THERMOPLASMA ACIDOPHILUM

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#### Summary

The obligate, thermophilic, acidophilic mycoplasma, Thermoplasma acidophilum, grows optimally at 56°C and pH 2.0. Its plasma membrane possessed 21-22 protein bands that were resolved by polyacrylamide gel electrophoresis. One major membrane protein, molecular weight 152 000, which stained for carbohydrate with periodic acid-Schiff reagent, accounted for 32% (w/w) of the total membrane proteins. It was isolated and further purified by concanavalin A affinity chromatography. The carbohydrate content amounted to less than 10% (w/w) compared to that of the entire glycoprotein. The carbohydrate moiety consisted mainly of mannose residues with branched  $\alpha$  1  $\rightarrow$  2 linkages at the non-reducing ends of the glycopeptide as determined by permethylation followed by gas chromatography-mass spectrometry analysis. The reducing end was an N-glycosidic linkage between asparagine and N-acetylglucosamine. The amino acid composition of this glycoprotein showed 62 mol% hydrophobic residues, while the acidic amino acid content contributed 9 mol% more than that of the basic amino acids. The existence of membrane glycoproteins in the procaryotic, wall-less T. acidophilum may provide a protective coat for the plasma membrane. The stereochemistry and the conformation of the carbohydrate chains, in conjunction with water turgor, may contribute to the rigidity of the membrane and the cation binding.

#### Introduction

One of the major modifications that a protein may undergo after synthesis of its peptide chains is the attachment of sugar residues. A large number of

proteins of diverse origin and biological function are known to contain such covalently linked carbohydrates and are called glycoproteins. Glycoproteins are widely distributed in eucaryotes, occurring not only in vertebrate and invertebrate animals, but also in plants and unicellular organisms. In particular, carbohydrate components of plasma membranes, which separate the cell from its external environment, appear to play a crucial role in active transport of molecules, serve as receptors for viruses, hormones, and antibodies, and take part in intercellular recognition and adhesion [1].

Glycoproteins, though quite common in eucaryotes, are rare in procaryotes [2,3], and the search for glycoproteins in mycoplasma membranes is still in its initial stage [4]. A thermophilic mycoplasma, viz. Thermoplasma acidophilum, grows optimally at pH 2 and 56°C [5]. The organism is totally devoid of a cell wall, only its plasma membrane encounters the harsh environment. Besides structurally unique glycerol ether membrane lipids [6,7], we discovered the existence of membrane glycoproteins [8] which may play a role in the survival of T. acidophilum. This primitive microorganism may serve as a model to explore the complex interactions between cell and its extreme environment. In this communication, we are specifically concerned with the elucidation of structural and functional relationships of membrane glycoproteins. The experimental results will also provide a better comprehension of the mechanism of resistance of cells that are exposed to environmental stress.

#### Materials and Methods

## Growth of cells

T. acidophilum was originally obtained from the American Type Culture Collection (ATCC 25905). Cells were grown and harvested as described previously [9].

# Membrane preparation

Cells were lysed by 1 M glycine buffer, pH 9.3, 22°C, at a protein concentration higher than 10 mg/ml. Membranes were collected and purified on a discontinuous sucrose gradient, followed by three water washes. The purity of the membrane preparation was ascertained by electron microscopy, absence of DNA, and absence of cytoplasmic contaminants [9].

# Buffer system for SDS gel electrophoresis and sample preparation

The discontinuous SDS buffer system of Laemmli [10] was used. Separating gels usually contained 8% acrylamide, while stacking gels contained 5% acrylamide; they were prepared from the same stock solution. The final concentration of SDS was 0.1% in both kinds of gel and in the electrode buffer. Sample preparation was performed as described [10], except that samples were boiled for 2 min.

#### Isolation and purification of membrane glycoprotein

By preparative slab gel electrophoresis. The buffer system of Laemmli [10] was used to run slab gels. The slab gel apparatus (Bio-Rad, Richmond, CA) was operated at 30 mA with a slab thickness of 3 mm. After electrophoresis, three

strips of gels were cut vertically and stained with periodic acid-Schiff and Coomassie blue. The gel was reconstructed after staining and glycoprotein bands were cut, and eluted with buffer. SDS was removed by dialysis.

By phenol extraction. Membranes were extracted with 50% phenol solution at 70°C for 15 min. Material at the interface was collected and dialysed.

By Sepharose 4B column chromatography. Membranes were solubilized by 1% SDS and loaded onto a  $90\times1.5$  cm (inner diameter) Sepharose 4B (Pharmacia, Uppsala, Sweden) column. Fractions of 3 ml were collected and assayed for protein and carbohydrate.

By concanavalin A-Sepharose column chromatography. Glycoproteins isolated by the above three methods can be further purified by concanavalin A-Sepharose (Sigma, St. Louis, MO) column. Glycoprotein was loaded on the column and eluted first with a solution of 0.5% Triton X-100, 50 mM Tris, pH 8.0, 0.5 M NaCl, and then with 0.5% Triton X-100, 50 mM Tris, pH 8.0, 0.2 M  $\alpha$ -methylmannoside. Triton X-100 was removed by dialysis.

## Methanolysis of carbohydrates

Methanolysis of glycoprotein to study the carbohydrate portion was performed by the method of Chambers and Clamp [11]. Trimethylsilylated methyl glycosides were analyzed on a Perkin-Elmer gas chromatograph, Model 910, with a 12 ft, 3% SP-2100 column (80—100 mesh, on Supelcoport) with an initial hold of 4 min at 120°C, then programmed at 0.5°C/min to 185°C, and held at 185°C for 20 min.

# Permethylation for linkage studies

Permethylation of carbohydrates was performed by the method of Hakomori [12]. All the permethylation operations were carried out under dry nitrogen. After acid hydrolysis, borohydride reduction, and acetylation, the partially methylated alditol acetates in chloroform were ready for analyses by GLC or GLC-MS. Column packings used for these analyses were 3% OV-210 on Gas Chrom Q (80—100 mesh) or 3% OV-225 on Gas Chrom Q (80—100 mesh). The mass spectra were interpreted according to Björndal et al. [13].

## Anhydrous hydrogen fluoride deglycosylation

The reaction vessel containing freeze-dried sample was cooled in a liquid nitrogen bath. 10 ml of HF was distilled over from the reservoir. The reaction vessel was allowed to warm up to room temperature. At the end of the reaction, the reaction vessel was evacuated via a calcium oxide trap. To ensure complete removal of HF, the line was evacuated for an additional hour after removal of visible HF. The sample was then taken up in water and passed through an Amicon ultrafiltration cell to separate the carbohydrate and the protein [14].

#### Amino acid analysis

5.7 N HCl was added to the purified glycoprotein. Hydrolysis was complete after 18 h at 110°C. The suspension was taken to dryness under nitrogen and resuspended in 0.01 N HCl. Amino acid analysis was performed on a single column accelerated flow Technicon system modified by Dr. D.T.A. Lamport, Plant Research Laboratory, Michigan State University.

 $\alpha$ -Mannosidase digestion. Reaction mixtures containing approximately 5 mg of purified glycoprotein in 1 ml of 0.1 M citrate buffer, pH 4.5, were incubated at 37°C with 2–3 units of  $\alpha$ -mannosidase. Incubation was continued until no additional release of mannose was observed. The reaction was terminated by introducing 2 ml of 0.2 M borate buffer, pH 9.8. The reaction mixtures were then passed through an ultrafiltration cell to separate the free carbohydrates from the rest of the glycoproteins.

 $\beta$ -Glucosidase digestion. The same procedure for  $\alpha$ -mannosidase was used for  $\beta$ -glucosidase.

 $\beta$ -Galactosidase digestion. The same procedure for  $\alpha$ -mannosidase was used; only the buffer (0.05 M phosphate, pH 7) was different.  $\beta$ -Galactosidase has an optimum pH at 7.0. Mannosidase, glucosidase and galactosidase were purchased from Sigma Chemical Co. (St. Louis, MO).

Endoglycosidase H digestion. Glycopeptide in 1 ml of 0.1 M acetate buffer, pH 5.0, was incubated with one unit of endoglycosidase H for 16 h at  $37^{\circ}$ C. The reaction was stopped by heating for 1 min in a boiling water bath and the reaction mixture ultrafiltrated. Endoglycosidase H (Miles, Elkhart, IN) from Streptomyces griseus hydrolyzes the  $\beta$ -di-N-acetylchitobiose structure in asparagine-linked sugar chains of glycopeptides.

#### Results

SDS-polyacrylamide gel electrophoresis of membrane proteins showed 21–22 protein bands (Fig. 1). Two bands which have the apparent molecular weights of 180 000 and 152 000 (1/4 ratio, w/w) were periodic acid-Schiff positive (Fig. 1).

# Isolation and purification of the glycoprotein

Several methods were used to isolate the glycoproteins. The three most satisfactory ones are listed in Materials and Methods. Using slab gel electrophoresis, the quantities of the isolated glycoproteins are rather small. Since 1 mg of membrane proteins could be maximally loaded onto a slab to get good resolution, the glycoproteins recovered from the gel eluent were in the range of 200 µg. This was a rather small quantity for structural analysis. Therefore, phenol extraction was used for the first step in glycoprotein isolation. Glycoproteins were recovered at the interface between water and phenol. After dialysis to remove phenol, the glycoproteins were solubilized in 1% SDS, then loaded onto a 90 by 1.5 cm (inner diameter) Sepharose 4B column to separate glycoproteins. Further purification could be achieved by application to a concanavalin A-Sepharose column (Fig. 2). A Thermoplasma glycoprotein (apparent molecular weight 152 000) was isolated and purified by the procedure listed above (Fig. 3). The purified glycoprotein was homogeneous showing a single band by gel electrophoresis in three different concentrations of acrylamide (5%, 7%, and 9%); staining was performed with Coomassie blue and periodate-Schiff reagents.

Since many glycoproteins stain very poorly with Coomassie blue, it will be inaccurate to estimate the glycoprotein content in the membrane by staining intensity. Membrane proteins and isolated glycoproteins were determined by

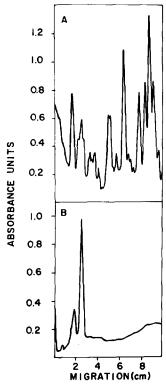


Fig. 1. Sodium dodecyl sulfate-polyacrylamide gel electrophoresis pattern of plasma membrane proteins from *T. acidophilum*. (A) The gel was stained with Coomassie blue and scanned at 550 nm. (B) The gel was stained with periodic acid-Schiff reagent and scanned at 540 nm.

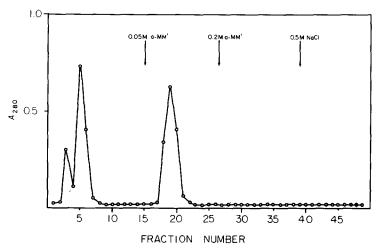


Fig. 2. Elution pattern of plasma membrane glycoprotein from concanavalin A-Sepharose column. Glycoprotein isolated from T. acidophilum membrane was loaded on the concanavalin A-Sepharose column (10 by 0.5 cm, inner diameter) and eluted first with a solution of 0.5% Triton X-100, 50 mM Tris, pH 8.0, 0.5 M NaCl, and then with 0.5% Triton X-100, 50 mM Tris, pH 8.0, 0.05 M  $\alpha$ -methylmannoside ( $\alpha$ -MM'), then with 0.5% Triton X-100, 50 mM Tris, pH 8.0, 0.2 M  $\alpha$ -MM', and finally with 0.5% Triton-X-100, 50 mM Tris, pH 8.0, 0.5 M NaCl. The arrows indicate changes of elution buffers.

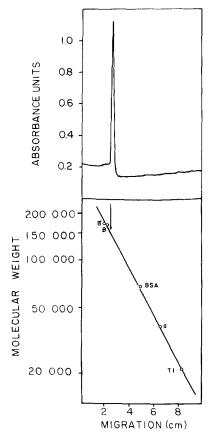


Fig. 3. Sodium dodecyl sulfate-polyacrylamide gel electrophoresis pattern of the purified glycoprotein from concanavalin A-Sepharose column and the molecular weight determination. The gels were stained with Coomassie blue. Molecular weight standards are:  $\beta'$ -subunit of RNA polymerase from Escherichia coli (165 000),  $\beta$ -subunit of RNA polymerase from E. coli (155 000), bovine serum albumin (BSA) (68 000),  $\alpha$ -subunit of RNA polymerase from E. coli (39 000) and trypsin inhibitor (TI) (21 500).

the method of Wang and Smith [15] with Triton X-100. The two membrane glycoproteins accounted for approximately 40% (w/w) of the total membrane proteins. Since the glycoprotein with an apparent molecular weight of 152 000 seems to be the major membrane glycoprotein, the following analysis deals with this membrane component.

# Pronase digestion of the glycoprotein

Proteolytic digestion of this purified glycoprotein was performed with pronase (Calbiochem) in 0.01 M Tris buffer, pH 7.8, at an enzyme-to-protein ratio of 1:10. Mixtures were incubated at 37°C for 48 h with 0.02% sodium azide added to prevent bacterial growth. After digestion, the sample was loaded onto a 90 by 1.5 cm (inner diameter) Sephadex G-100 column. The anthrone-positive fraction was then loaded onto a 90 by 1.5 cm (inner diameter) Sephadex G-75 column to give further purification of the glycopeptides. Gel electrophoresis of the purified glycopeptides showed a single band with an apparent molecular weight of 15 000. Molecular weight estimations by gel elec-

trophoresis of highly glycosylated proteins are known to give anomalously high values. 15 000 should be considered as the upper limit for the true molecular weight of the glycopeptides. This suggests that the carbohydrate content of the isolated glycoprotein amounts to 8–10%. This value was also confirmed colorimetrically [16], and the analytically determined carbohydrate content (Table I). Therefore, both pronase digestion followed by column chromatography and SDS gel electrophoresis indicated reproducibly a covalently linked glycopeptide. We concluded that the carbohydrates were not present as contaminants in the protein. The covalent linkage was further confirmed by methods described below.

# Determination of carbohydrate composition by methanolysis

Methanolysis of glycoprotein to study the carbohydrate portion was performed by the method of Chambers and Clamp [11]. The trimethylsilylderivatized methyl glycosides were run on a 12 ft, 3% SP-2100 column with a Perkin-Elmer gas chromatograph as described in Materials and Methods. A set of standards (mannose, glucose, galactose, glucosamine) and internal standards were used for quantitative calibration. Gas chromatographic analysis showed that the carbohydrate portion of the glycoprotein consisted mainly of mannose, glucose, and galactose (molar ratio 20/2/1) and a trace amount of glucosamine (Fig. 4). This finding was consistent with the concanavalin A binding ability of the isolated glycoprotein [17].

TABLE I

AMINO ACID AND CARBOHYDRATE COMPOSITION OF THE PURIFIED GLYCOPROTEIN ISOLATED FROM THERMOPLASMA ACIDOPHILUM

	$\mu mol/100 mg$	mol%	
Lysine	30.8	4.3	
Histidine	10.1	1.4	
Arginine	28.0	3.9	
Aspartic acid	80.4	11.2	
Threonine	58.9	8.2	
Serin <i>e</i>	73.9	10.3	
Glutamic acid	53.8	7.5	
Proline	7.9	1.1	
Glycine	68.9	9.6	
Alanine	58,1	8,1	
Valine	53.8	7.5	
Methionine	3.6	0.5	
Isoleucine	41.6	5.8	
Leucine	59.6	8.3	
Tyrosine	51.7	7.0	
Phenylalanine	35.9	5.0	
Mannose	52.1		
Glucose	5.1		
Galactose	2.5		
Glucosamine	1.3		

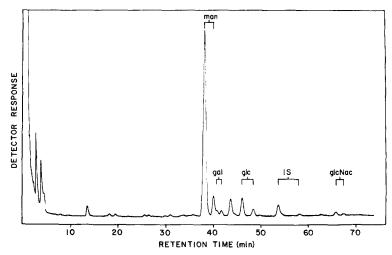


Fig. 4. Gas-liquid chromatography of trimethylsilylated methyl glycosides of the purified glycopeptide from the plasma membrane of *T. acidophilum*. Analysis was performed on a 12 ft, 3% SP-2100 column (80–100 mesh, on Supelcoport). Mannitol was used as internal standard (I.S.). The peak eluted at about 43 min was not identified.

## Amino acid analysis of the glycoprotein (Table I)

The amino acid composition of the glycoprotein was analyzed as described in Materials and Methods. Hydrophobic amino acids, as defined by Tanford [18], comprised at least 62 mol% of the protein portion, while acidic amino acids amounted to 9 mol% more than the basic amino acids.

# Glycosidic linkage studies of the glycoprotein

The purified glycoprotein resisted alkaline hydrolysis [19]; this suggested that the carbohydrates were N-glycosidically linked to the peptide.

Exposure of glycoproteins to anhydrous hydrogen fluoride cleaves all the linkages of neutral and acidic sugars while leaving peptide bonds and glycopeptide linkages of amino sugars intact [14]. Following HF treatment, the protein portion was separated from the carbohydrate moiety by dialysis. Methanolysis was carried out on the protein portion. Glucosamine was the only sugar detected.

# Digestions of glycopeptide by exo- and endoglycosidases

The purified glycoprotein was first incubated with  $\alpha$ -mannosidase. This enzyme was selected since the glycoprotein can be purified by concanavalin A affinity chromatography. Also from the carbohydrate content (Fig. 4), mannose accounted for 85%. After 72 h of incubation (Materials and Methods), 70% (w/w) of the mannose was released. Further digestion with  $\alpha$ -mannosidase in the presence of SDS (1%) for another 48 h did not remove any more mannose. The glycopeptide recovered from  $\alpha$ -mannosidase digestion was further treated with  $\beta$ -glucosidase and  $\alpha$ -mannosidase. This reaction enabled the release of 62% of the glucose and 90% of the mannose residues (Table II). Further digestion with  $\beta$ -galactosidase,  $\beta$ -glucosidase and  $\alpha$ -mannosidase resulted in virtually total removal of the galactose and glucose residues

TABLE II

MOLAR RATIOS OF CARBOHYDRATE RESIDUES FROM PURIFIED GLYCOPROTEIN AND AFTER TREATMENTS WITH GLYCOSIDASE

Molar ratios were calculated by setting the value for glucosamine as 1.0. The glucosamine content remained constant after prolonged hydrolysis. The values in parentheses were calculated as mol carbohydrate/glycoprotein.

	Glycoprotein	After digestion with:			
		α-Mannosidase	lpha-Mannosidase $eta$ -glucosidase	lpha-Mannosidase $eta$ -glucosidase $eta$ -galactosidase	α-Mannosidase β-glucosidase β-galactosidase endoglycosidase
Mannose	40.1 (80.2)	11.8	3.9	0.9	_
Glucose	3.9 (7.8)	3.9	1.5	trace	_
Galactose	1.9 (3.8)	1.9	1.9	trace	-
Glucosamine	1.0 (2.0)	1.0	1.0	1.0	(1.0)

(Table II). From the experiments on exoglycosidase digestions, it was concluded that the carbohydrate portion of the glycoprotein had a long chain of  $\alpha$ -mannose (55–56 residues) at its non-reducing end, followed by a region containing six residues of  $\beta$ -glucose, and 16 of  $\alpha$ -mannose, followed by another region containing four residues of  $\beta$ -galactose, three residues of  $\beta$ -glucose, and six residues of  $\alpha$ -mannose. After the exoglycosidase digestion, the glycopeptide was treated with endoglycosidase H. Since this enzyme removed all the remaining carbohydrates except one glucosamine residue (Table II), this indicated that the reducing end comprised two N-acetylglucosamine residues (chitobiose),  $\beta$ -linked, followed by mannose (1–2 residues). The inability to remove the last trace (1–2 residues) of mannose with  $\alpha$ -mannosidase, suggested that at least one of the mannose residues was  $\beta$ -linked.

The glycosidases were used without further purification. Nevertheless, results from permethylation experiments, followed by mass spectrometry, supported all the information derived from studies with glycosidases.

#### Permethylation studies of the glycopeptides

To determine linkages, glycoproteins were first treated with pronase and glycosidases, then subjected to permethylation [12]. Partially methylated acetates were analyzed by combined gas-liquid chromatography-mass spectrometry and the data were interpreted as described [13]. Partially methylated derivatives of mannose, glucose, galactose, and glucosamine were identifed both by their retention times in the gas chromatograms and by mass spectrometry by their fragmentation patterns, and were further compared with known reference standards. Permethylation and mass spectral analyses of the glycopeptides gave evidence for the presence of 2,3,4,6-tetra-O-methylmannitol, 3,4,6-tri-O-methylmannitol, 2,4,6-tri-O-methylmannitol, 2,3,6-tri-O-methylglucitol, 2,4,6-tri-O-methylglactitol, and 3,6-di-O-methylglucosaminitol (Table III). The presence of 2,3,4,6-tetra-O-methylmannitol as the only tetramethylated carbohydrate, indicated that there were eight non-reducing terminal mannoses. The presence of an abundant

#### TABLE III

MOLAR RATIOS AND RETENTION TIMES OF PARTIALLY METHYLATED ALDITOL ACETATES OF MANNOSE, GLUCOSE, GALACTOSE AND GLUCOSAMINE OBTAINED BY PERMETHYLATION OF THE PURIFIED GLYCOPEPTIDE

Molar ratios were calculated using 1,4,5-tri-O-acetyl-3,6-di-O-methylglucosaminitol as 1.0. The values in parentheses were calculated as mol carbohydrate/glycoprotein. Retention time analysis was performed on a 6 ft, 3%, OV-225 (80—100 mesh on Sypelcoport) and operated isothermally for 50 min at 170°C, then programmed from 170°C to 250°C at 5°C/min, with helium as carrier gas. Quantitative evaluation of the chromatograms was performed with a Hewlett-Packard integrator, model 3370B.

	Molar ratios		Retention time (min)
	Before $lpha$ -mannosidase digestion	After $\alpha$ -mannosidase digestion	· <b>=</b> ,
1,5-Di-O-acetyl-2,3,4,6-tetra-O-methylmannitol	4.1 (8.2)	0	12,2
1,2,5-Tri-O-acetyl-3,4,6-tri-O-methylmannitol	18.4 (36.8)	trace	22.3
1,3,5-Tri-O-acetyl-2,4,6-tri-O-methylmannitol	6.1 (12.2)	6.3	23.2
1,5,6-Tri-O-acetyl-2,3,4-tri-O-methylmannitol	5.1 (10.2)	5.1	26.7
1,3,5,6-Tetra-O-acetyl-2,4-di-O-methylmannitol	3.5 (7.0)	trace	52.1
1,4,5-Tri-O-acetyl-2,3,6,-tri-O-methylglucitol	3.1 (6.2)	2.4	28.3
1,3,5-Tri-O-acetyl-2,4,6-tri-O-methylgalactitol	1.5 (3.0)	1.7	24.7
N-Acetyl-N-methyl-1,4,5-tri-O-acetyl-3,6-di-O- methylglucosaminitol	1.0 (2.0)	1.0	61.4

amount of 3,4,6-tri-O-methylmannitols (1,2-substituted) was consistent with the glycoprotein's capacity to bind concanavalin A. The presence of 2,4-di-O-methylmannitol (1,3,6-substituted) was indicative of a branched glycopeptide. From the molar ratios (Table III), it was obvious that the glycopeptide branched at seven locations. 2,4,6-Tri-O-methylmannitol (1,3-substituted) and 2,3,4-tri-O-methylmannitol (1,6-substituted) seemed to occur in about equal amounts. After  $\alpha$ -mannosidase digestion, the 2,4-di-O-methylmannitol and 3,4,6-tri-O-methylmannitol derivatives were drastically reduced (Table III). This indicated that the branching was entirely at the outer mannose residues which amounted to 70% (w/w) (Table II) and which were linked predominantly via  $\alpha$  1  $\rightarrow$  2 linkages. The middle inner core of the carbohydrate moiety

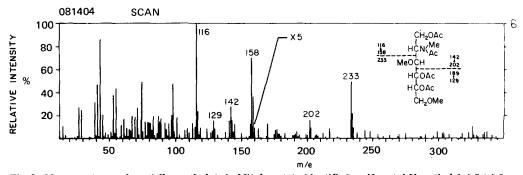
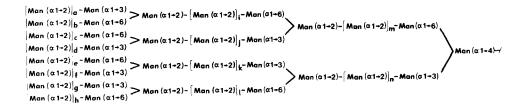


Fig. 5. Mass spectrum of partially methylated alditol acetate identified as N-acetyl-N-methyl-1,4,5-tri-O-acetyl-3,6-di-O-methylglucosaminitol from the purified glycopeptide of the T. acidophilum plasma membrane.



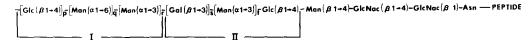


Fig. 6. Proposed structure of the T. acidophilum plasma membrane glycoprotein, a+b+c+d+e+f+g+h+i+j+k+l+m+n=39 residues; q+r+t=20 residues; p=1 five residues; p=1 five residues; p=1 five residues. The sequence of the residues in segment I and II is not certain.

consisted of 1,4-substituted glucose, 1,3-substituted galactose, 1,2-, 1,3-, and 1,6-substituted mannose. The innermost section of the carbohydrate region comprised 1,4-substituted glucosamine residues (Fig. 5). Because  $\alpha$ -mannosidase (Table II) was incapable of removing the last trace of mannose, this suggested to us that at least one mannose residue was  $\beta$ -linked to the distal N-acetyl-glucosamine residue. A proposed structure of the glycopeptide is shown in Fig. 6.

#### Discussion

The purified membrane glycoprotein from T. acidophilum had an apparent molecular weight of 152 000 with less than 10% carbohydrate content (w/w). The carbohydrate moiety consisted mainly of mannose residues with branched  $\alpha 1 \rightarrow 2$  linkages at the non-reducing ends of the glycopeptide. The reducing end probably was an N-glycosidic linkage between the amino acid asparagine and N-acetylglucosamine, as evidenced by resistance to alkaline hydrolysis, cleavage by anhydrous HF, and endoglycosidase reactions. At least one mannose residue was not  $\alpha$ -linked. The glycoprotein accounted for 40% (w/w) of the total membrane proteins. The non-reducing ends of the glycopeptide showed a highly branched pattern (Fig. 6) which may extend over the entire cell surface. This was supported by electron microscopic studies on Thermoplasma membranes labelled with concanavalin A [20].

N-Glycosidically linked, mannose-rich glycoproteins have been found in animal systems [21], fungi [22], yeast [22], and plants [23]. It appears that we have demonstrated for the first time a N-glycosidically linked, mannose-rich procaryotic glycoprotein. Preliminary experiments carried out in our laboratory appear to indicate that glycosylation of certain membrane proteins may contribute to the survival ability of T. acidophilum at extreme environments. We supplied growing cultures of T. acidophilum with bacitracin. There seems to exist a correlation of growth at increasing pH values and the reduced amount of carbohydrates in membrane glycoproteins. Bacitracin is an anti-

biotic which blocks peptidoglycan biosynthesis in bacteria by complexing with the lipid pyrophosphate released after transfer of the lipid-linked subunit to the growing peptidoglycan chain [24].

The amino acid composition of the glycoprotein showed 62 mol% hydrophobic residues, while the acidic amino acid content contributed 9 mol% more than that of the basic amino acids. The amount of chargeable groups (at most 28 mol%) may be diminished further at the low optimal pH value because of the protonation of free carboxyl groups which predominate over the number of free amino groups. A high degree of hydrophobicity is of survival value since it possesses advantages for thermophily as well as acidophily. Hydrophobic interactions have a maximum stability around 60°C [25] which is very close to the optimum growth temperature of *T. acidophilum*.

The native structure of proteins is stabilized by a combination of hydrogen bonding, electrostatic and hydrophobic interactions [26].

In the *Thermoplasma* membrane, 40% of the proteins consist of a mannoserich glycoprotein. As far as interactions between the carbohydrate moiety and the membrane proteins are concerned, hydrophobic interactions probably play an important role [26]. Consequently the effect of carbohydrates on proteins depends on the influence of sugars upon water structure. Constructing Pauling-Corey spacefilling models of the carbohydrate portion of the *Thermoplasma* glycoprotein, one easily visualizes the extreme flexibility of the 1,6-linkage between the corresponding [27] carbohydrate sections. Furthermore, sugar conformations may be visualized capable of generating polysaccharide surfaces which are virtually hydrophobic. The mutual hydroxyl orientations play obviously a considerable role. The array of hydroxyl groups participates in the formation of hydrogen bonds with water.

Interactions between regular periodic mannose portions of glycoproteins may generate periodic conformations such as a helix or a ribbon [28]. Thus it is conceivable that the *Thermoplasma* surface is surrounded by a mesh-like carbohydrate coat where water molecules are immobilized. The resulting water turgor and the interactions between carbohydrates provide perhaps additional factors for the rigidity of the limiting plasma membrane. Indeed, electron paramagnetic resonance studies [29] have shown that the *T. acidophilum* membrane is the most rigid membrane reported hitherto.

The capacity of neutral or anionic carbohydrates [28] to form complexes may further contribute to the stability of the membrane. Given the appropriate coordination geometry, cations may link carbohydrates with the underlying protein-lipid framework. The rigidity of the *Thermoplasma* membrane is actually enhanced in the presence of Ca<sup>2+</sup> or Al<sup>3+</sup> [30,31]. It appears worth while to mention that rather limited information is available regarding structural and thermodynamic properties of 1,2-linked polysaccharides [32].

In conclusion, the existence of a mannose-rich glycoprotein in the wall-less *T. acidophilum* provides presumably a protective coat towards the harsh environment. The stereochemistry and conformation of the carbohydrate network, in conjunction with an ensuing hydroskeleton, may contribute to the stability of the membrane and the cation binding.

## Acknowledgements

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#### References

- 1 Kornfeld, R. and Kornfeld, S. (1976) Annu. Rev. Biochem. 45, 217-237
- 2 Razin, S. (1975) Prog. Surf. Membrane Sci. 9, 257-312
- 3 Mescher, M.F. and Strominger, J.L. (1976) Proc. Natl. Acad. Sci. U.S. 73, 2687-2691
- 4 Razin, S. (1978) Microbiol. Rev. 42, 414-470
- 5 Darland, G., Brock, T., Samsonoff, W. and Conti, S.F. (1970) Science 170, 1416-1418
- 6 Yang, L.L. and Haug, A. (1977) Biophys. J. 17, 96a
- 7 Langworthy, T.A. (1977) Biochim. Biophys. Acta 487, 37-50
- 8 Yang, L.L. and Haug, A. (1975) Biophys. J. 21, 206a
- 9 Ruwart, M.J. and Haug, A. (1975) Biochemistry 14, 860-866
- 10 Laemmli, U.K. (1970) Nature 277, 680-685
- 11 Chambers, R.E. and Clamp, J.R. (1971) Biochem. J. 125, 1009-1018
- 12 Hakomori, S.-I. (1964) J. Biochem. (Tokyo) 55, 205-208
- 13 Björndal, H., Hellerqvist, C.G., Lindberg, B. and Svensson, S. (1970) Angew. Chem. Int. Edn. 8, 610-619
- 14 Mort, A.J. and Lamport, D.T.A. (1977) Anal. Chem. 82, 289-309
- 15 Wang, C.-S. and Smith, R.L. (1975) Anal. Biochem, 63, 414-417
- 16 Roe, J.H. (1955) J. Biol, Chem. 212, 335-343
- 17 Goldstein, I.J. (1976) in Concanavalin A as a Tool (Bittiger, H. and Schnebli, H.P., eds.), pp. 55-57, Wiley Publ., New York
- 18 Tanford, C. (1973) in The Hydrophobic Effect, pp. 120-121, Wiley Publ., New York
- 19 Spiro, R.G. (1971) Methods Enzymol, 28B, 3-43
- 20 Mayberry-Carson, K.J., Jewell, M.J. and Smith, P.P. (1978) J. Bacteriol. 133, 1510-1513
- 21 Montreuil, J. (1975) Pure Appl. Chem. 42, 431-477
- 22 Lucas, J.J., Waechter, C.J. and Lennarz, W.J. (1976) J. Biol. Chem. 250, 1992-2002
- 23 Lis, H. and Sharon, N. (1978) J. Biol. Chem. 253, 3468-3476
- 24 Storm, D.R. and Strominger, J.L. (1973) J. Biol. Chem. 248, 3940-3945
- 25 Scheraga, H. (1963) in The Proteins (Neurath, H., ed.), pp. 477-594, Academic Press, New York
- 26 Lakshmi, T.S. and Nandi, P.K. (1976) J. Phys. Chem. 80, 249-252
- 27 Rees, D.A. (1972) Biochem. J. 126, 257-273
- 28 Rees, D.A. (1975) in Biochemistry of Carbohydrates, MTP International Review of Science (Whelan, W.J., ed.), Vol. 5, pp. 1-42, University Park Press, Baltimore, MD
- 29 Smith, G.G., Ruwart, M.J. and Haug, A. (1974) FEBS Lett. 45, 96-98
- 30 Weller, H.G. and Haug, A. (1977) J. Gen. Microbiol. 99, 379-382
- 31 Vierstra, R. and Haug, A. (1978) Biochem. Biophys. Res. Commun. 84, 138-143
- 32 Suggett, A. (1975) in Water (Franks, F., ed.), Vol. 4, pp. 519-567, Plenum Press, New York